

## Ionic Bonding Chemistry If8766 Answers

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Ionic bonding When metals react with non-metals, electrons are transferred from the metal atoms to the non-metal atoms, forming ions . The resulting compound is called an ionic compound .

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IONIC BONDING Name Ionic bonding occurs when a metal transfers one or more electrons to a nonmetal in an effort to attain a stable octet of electrons, For example, the transfer of an electron from sodium to chlorine can be depicted by a Lewis dot diagram. + Calcium would need two chlorine atoms to get rid of its two valence electrons.

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Chapter Organic Chemistry Existence Summary We Tried to Identify Some Great Ionic and Covalent Bonding Worksheet Answer Key Together with Chemistry If8766 Answer Key Photo A picture to suit your needs. Metals create positively charged ions rather than metal creating negatively charged ions. Ionic compounds

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Ionic bonding results in compounds known as ionic, or electrovalent, compounds, which are best exemplified by the compounds formed between nonmetals and the alkali and alkaline-earth metals. In ionic crystalline solids of this kind, the electrostatic forces of attraction between opposite charges and repulsion between similar charges orient the ions in such a manner that every positive ion ...

~~ionic bond | Definition, Properties, Examples, & Facts ...~~

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